

# MISSOURI DEPARTMENT OF TRANSPORTATION MATERIALS ENGINEERING Jefferson City, Missouri

## Test Method MoDOT T28 DETERMINATION OF POTASH IN FERTILIZER

#### 1.0 SCOPE

**1.1** This method describes a procedure for determining the percent Potash in Fertilizers by Atomic Absorption Spectrophotometry.

#### 2.0 REAGENTS AND APPARATUS

- **2.1** Atomic Absorption Spectrophotometer conforming to the specifications set forth in ASTM C 114.
- **2.2** Potassium Chloride KCl, Reagent Grade, dried at 105-110 C for several hours prior to use.
- 2.3 Lanthanum Oxide La<sub>2</sub>0<sub>3</sub>, Reagent Grade.
- **2.4** Hydrochloric Acid HCl, Sp. Gr. 1.19.

#### 3.0 PREPARATION OF STANDARD SOLUTIONS

- 3.1 Lanthanum stock solution: Weigh 23.46 gr. of La<sub>2</sub>0<sub>3</sub> into a 600 ml beaker and add 400 ml H<sub>2</sub>O. While stirring add 40 ml HCl. Warm and stir on a magnetic stirrer until solution is complete. Cool to room temperature, filter into a 1000 ml volumetric flask, and dilute to volume.
- 3.2 KCl standard solution: Weigh 2.5000 gr. KCl into a 250 ml volumetric flask. Add approximately 150 ml  $\rm H_2O$  and boil for 30 minutes. Cool and dilute to volume. Pipette 50 ml of this solution into a 500 ml volumetric flask, add 50.00 ml of the Lanthanum stock solution, and dilute to volume. This solution is 63.18%  $\rm K_2O$ .
- **3.3** Blank solution: Pipette 50 ml of the Lanthanum stock solution into a 500 ml volumetric flask and dilute to volume. This solution is used to set the zero point on the instrument.



#### 4.0 PROCEDURE

- **4.1** Weigh 2.4900 to 2.5000 gr. of the sample into a 250 ml volumetric flask, add approximately 150 ml H<sub>2</sub>O and boil 30 minutes. Cool to room temperature, dilute to volume, and let stand approximately 3 hours. Transfer a 10.00 ml aliquot to a 100 ml volumetric flask, add 10.00 ml of the Lanthanum stock solution and dilute to volume.
- **4.2** Calibrate the instrument using the Blank solution and the KCl standard solution, then determine the concentration of the sample solution.

### 5.0 CALCULATIONS

- **5.1** The method of calculating the %  $K_20$  will vary according to the make and model of instrument used.
- **5.2** Report the results, to the nearest 0.1%, as follows: % Potash ( $K_20$ )